**High School Science Tested Indicators** 

Indicator#	Text of Indicator	Physical or Life Science Test Component
	▲ actively engages in investigations, including developing	Component
S.HS.1.1.2 S.HS.1.1.3		
	questions, gathering and analyzing data, and designing	Р
	and conducting research  A actively engages in using technological tools and	
		Р
S.HS.2A.1.1	mathematics in their own scientific investigations.	
	▲ understands atoms, the fundamental organizational	
	unit of matter, are composed of subatomic particles.	
	Chemists are primarily interested in the protons,	P
0.110.24.2.1	electrons, and neutrons found in the atom.	P
S.HS.2A.2.1	▲ understands chemists use kinetic and potential energy	
	to explain the physical and chemical properties of matter	
	on earth that may exist in any of these three states: solids,	
	liquids, and gases.	Р
S.HS.2A.2.2	▲ understands the periodic table lists elements according	
	to increasing atomic number. This table organizes	
	physical and chemical trends by groups, periods, and sub-	-
	categories.	Р
S.HS.2A.2.3	▲ understands chemical bonds result when valence	
	electrons are transferred or shared between atoms.	
	Breaking a chemical bond requires energy. Formation of a	
	chemical bond releases energy. Ionic compounds result	
	from atoms transferring electrons. Molecular compounds	
	result from atoms sharing electrons. For example, carbon	
	atoms can bond to each other in chains, rings, and	
	branching networks. Branched network and metallic solids	963
	also result from bonding.	Р
S.HS.2A.3.1	▲ understands a chemical reaction occurs when one or	
	more substances (reactants) react to form a different	
	chemical substance(s) (products). There are different	
	types of chemical reactions all of which demonstrate the	
	Law of Conservation of Matter and Energy.	
	20 25 2 20 MINER 10 M	Р
S.HS.2B.1.1	▲understands Newton's Laws and the variables of time,	
	position, velocity, and acceleration can be used to	
	describe the position and motion of particles.	Р
S.HS.2B.2.2	▲understands the first law of thermodynamics states the	
	total internal energy of a substance (the sum of all the	
	kinetic and potential energies of its constituent molecules)	
	will change only if heat is exchanged with the environment	
	or work is done on or by the substance. In any physical	
	interaction, the total energy in the universe is conserved.	
	1776	Р
S.HS.2B.3.2	▲understands waves have energy and can transfer	11090
	energy when they interact with matter.	Р
S.HS.2B.3.5	▲ understands electromagnetic waves resulty when a	5742
	charged particle is accelerated or decelerated.	Р
S.HS.3.1.2	▲ understands cell functions involve specific chemical	
	reactions.	L
S.HS.3.2.1	▲ understands living organisms contain DNA or RNA as	
	their genetic material, which provides the instructions that	
	specify the characteristics of organisms.	
		47

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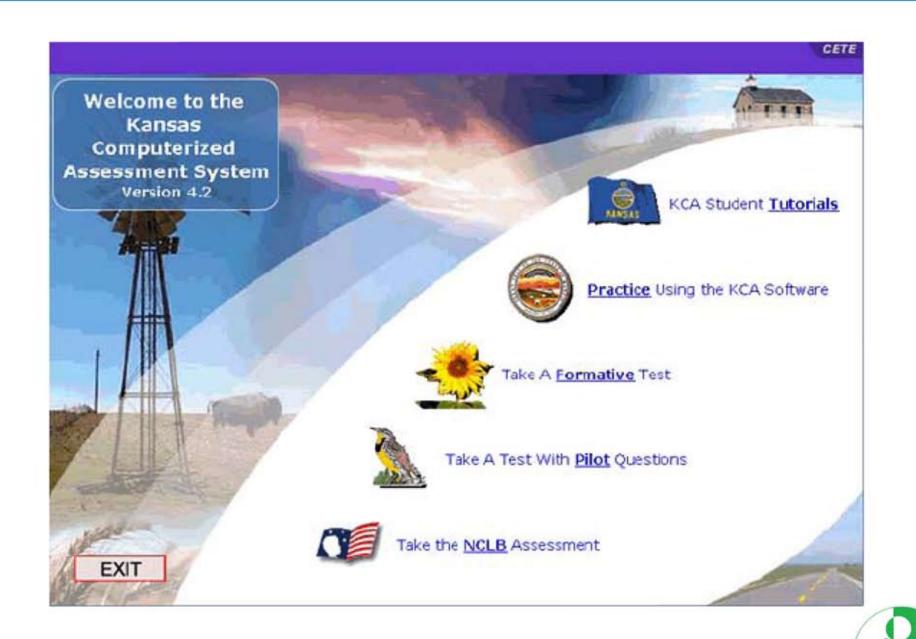
# **Elementary Science Tested Indicators**

Indicator #	Text of Indicator	
S.4.1.1.1	▲ asks questions that he/she can answer by investigating	
S.4.1.1.2	▲ plans and conducts a simple investigation	
S.4.1.1.3	▲ employs appropriate equipment, tools, and safety procedures to gather data	
S.4.1.1.4	▲ begins developing the abilities to communicate, critique, analyze his/her own investigations, and interpre- the work of other students	
S.4.2.1.1	▲ observes properties of objects and measures those properties using appropriate tools	
S.4.2.1.2	▲ describes and <i>classifies</i> objects by more than one property	
S.4.2.1.3	▲ observes and records how one object interacts with another object	
S.4.2.1.4	▲ recognizes and describes the differences between solids, liquids, and gases	
S.4.2.2.1 (This		
item not tested	▲ moves objects by pushing, pulling, throwing, spinning,	
on KAMM)	dropping, and rolling; and describes the motion	
S.4.2.3.1	▲ identifies that the source of sound is vibrations	
S.4.2.4.1	▲ demonstrates that magnets attract and repel	
S.4.2.4.3	▲ constructs a simple circuit	
S.4.3.1.1	▲ observes different organisms and compares and contrasts how similar functions are served by different structural characteristics	
S.4.3.1.2	▲ compares basic needs of different organisms in their environment	
S.4.3.2.1	▲ compares, contrasts, and asks questions about life cycles of various organisms	
S.4.4.1.1	▲ collects, observes properties, and classifies a variety of earth materials in his/her environment	
8.4.4.1.3	▲ describes properties of water and process of the water cycle	
S.4.4.2.3	▲ discusses that the sun provides light and heat (electro- magnetic radiation) to maintain the temperature of the earth	
S.4.4.3.1	▲ describes changes in the surface of the earth	
S.4.4.3.2	▲ observes, describes, and records daily and seasonal weather changes	
S.4.5.1.1	▲ identifies a simple design problem (designs a plan, implements the plan, evaluates the results, makes changes to improve the product, and communicates the results)	
S.4.6.1.1	▲ discusses the nutritional value of various foods and their contribution to health	

#### 7th Grade Science Tested Indicators

Indicator #	Text of Indicator
7.1.1.1	▲ identifies questions that can be answered through
1101011	scientific investigations.
7.1.1.2	▲ designs and conducts scientific investigations safely
	using appropriate tools, mathematics, technology, and
	techniques to gather, analyze, and interpret data.
	learning act to gainer, analyze, and interpret data.
7.1.1.3	▲ identifies the relationship between evidence and
-visitante	logical conclusions.
7.1.1.4	▲ communicates scientific procedures, results and
	explanations.
7.1.3.2	▲ evaluates the work of others to determine evidence
	which scientifically supports or contradicts the results,
	identifying faulty reasoning or conclusions that go
	beyond evidence and/or are not supported by data.
7.2.1.1	▲ compares and classifies the states of matter; solids,
	liquids, gases, and plasma
7.2.2.1	▲ understands the relationship of atoms to elements
	and elements to compounds.
7.2.2.2	▲ measures and graphs the effects of temperature on
	matter.
7.2.3.2	▲ describes, measures, and represents data on a graph
7121012	showing the motion of an object (position, direction of
	motion, speed).
7.2.3.3	▲ recognizes and describes examples of Newton's
/ IMIOIO	Laws of Motion.
7.2.3.4	▲ investigates and explains how simple machines
1.2.3.4	multiply force at the expense of distance.
7.2.4.2	▲ understands that when work is done energy
1.2.4.2	transforms from one form to another, including
	mechanical, heat, light, sound, electrical, chemical, and
	nuclear energy, yet is conserved.
7.2.4.3	▲ observes and communicates how light
1.2.4.3	(electromagnetic) energy interacts with matter:
	transmitted, reflected, refracted, and absorbed.
7.2.4.4	transmitted, renected, renacted, and absorbed.
/.2.4.4	A understands that heat operay can be transferred from
	▲ understands that heat energy can be transferred from hot to cold by radiation, convection, and conduction.
7.3.1.1	▲ will understand the cell theory; that all organisms are
	composed of one or more cells, cells are the basic unit
	of life, and that cells come from other cells.
7.3.1.2	▲ relates the structure of cells, organs, tissues, organ
	systems, and whole organisms to their functions
7.3.2.1	▲ differentiates between asexual and sexual
	reproduction of organisms.
7.3.3.1	▲ understands that internal and/or environmental
ACT CONC.	conditions affect an organism's behavior and/or
	response in order to maintain and regulate stable
	internal conditions to survive in a continually changing
	environment.
7.3.4.1	
· ············	▲ recognizes that all populations living together (biotic
	resources) and the physical factors (abiotic resources)
	with which they interact compose an ecosystem.

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Science

## **Science Module 5 Questions**

#### 1. True or False?

All high school students must take the physical science and life science halves of the assessment during their 10<sup>th</sup> grade year.

False

### 2. True or False?

If our 9<sup>th</sup> graders don't meet standard on part of the science assessment, they can retest in 10<sup>th</sup> grade.

False

#### 3. True or False?

If my 4<sup>th</sup> grader is taking the KAMM, he can be assessed over 3<sup>rd</sup> grade indicators.

False

